Molecular Rotations of Polyhydroxycyclohexanes in Relation to their Structures. II^{13} . (-)-1, 2, 3, 5/4-Cyclohexanepentol

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 $[M]_2^{20}(W)$ of (-)-1, 2, 3, 5/4-cyclohexanepentol ((-)-epi-quercitol²⁾) was calculated by using PM-method and Whiffen's method³⁾, respectively. The orientation of the unit groups, (OH). in the molecule is shown in Table I (Fig. 1).

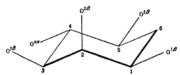


Fig. 1. Perspective drawing of the molecular model of (-)-1,2,3,5/4-cyclohexanepentol.

Table I

Name Orientation of unit group, (OH)

Cyclohexane

$$(-)$$
-1,2,3,5/ $[(1\beta),(2\beta),(3\beta), -6.8 -6.8 -6.8 (-10.9)^5)$
 $(-10.9)^5$

Calcd.

 $\sum [\mu]_{20\text{obs}}^{20}$ Whiffen (-3)

PM-Method.— [M] $_{\rm D}^{20}$ (W) of (-)-1, 2, 3, 5/4-cyclohexanepentol $\equiv \sum [\mu]_{\rm Dobs}^{20}$ of (-)-1, 2, 3, 5/4-cyclohexanepentol $= (1\beta) \, \mathbf{X} \, (2\beta) + (1\beta) \, \mathbf{X} \, (3\beta) + (1\beta) \, \mathbf{X} \, (4\alpha) + (1\beta) \, \mathbf{X} \, (5\beta) + (2\beta) \, \mathbf{X} \, (3\beta) + (2\beta) \, \mathbf{X} \, (4\alpha) + (2\beta) \, \mathbf{X} \, (5\beta) + (3\beta) \, \mathbf{X} \, (4\alpha) + (3\beta) \, \mathbf{X} \, (5\beta) + (4\alpha) \, \mathbf{X} \, (5\beta) = A + 0 + 0 + 0 - A - B + 0 - A + 0 + A^{40} = -B = -6.8 \, (-10.9)^{5}.$

¹⁾ Part I: S. Yamana, This Bulletin, 33, 1741 (1960).
2) "Advances in Carbohydrate Chemistry", Vol. 14, Academic Press, New York (1959), p. 190.

³⁾ D. H. Whiffen, Chem. & Ind., 1956, 964.

⁴⁾ Table II in the previous paper1) is used, here.

⁵⁾ The values, calculated by supposing that the optical center of (OH) is located at its O atom, are given in parentheses.

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Whiffen's Method.

along C^1-C^2 C|C-C|O+O|O-O|H+H|H-H|Calong C^2-C^3 C|H-C|C+O|C-O|O+H|O-H|Halong C^3-C^4 C|C-C|H+O|H-O|O+H|O-H|Calong C^4-C^5 C|H-C|C+O|O-O|H+H|C-H|Oalong C^5-C^6 C|C-C|H+O|H-O|H+H|H-H|Calong C^6-C^1 C|H-C|C+H|C-H|O+H|O-H|HC|H-C|C+H|C-H|O+H|O-H|H

As seen above, $[M]_{0}^{10}(W)$ of (-)-1, 2, 3, 5/4-cyclohexanepentol is presumed to be -6.8

 $(-10.9)^{5}$ (by PM-method) or 0 (by Whiffen's method). The corresponding observed value has not been found in any referential literatures, but $[M]_D(W)$ was observed at $27^{\circ}C$ as -8.9^{6} . This is nearly equal to $-6.8 (-10.9)^{5}$, the value calculated by PM-method at $20^{\circ}C$. This fact indicates that in this case also, PM-method is more suitable for estimating $[M]_D^{20}$ (W) than Whiffen's method.

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⁶⁾ B. Magasanik, R. E. Franzl and E. Chargaff, J. Am. Chem. Soc., 74, 2618 (1952).